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A.  $1.20 \times 10^3$  B.  $3.01 \times 10^3$  C.  $6.02 \times 10^3$  D.  $1.20 \times 10^4$  2. 1 mol of a hydrocarbon with general formula  $C_nH_{2n+2}$  reacts completely with oxygen to produce 4 mol of  $H_2O$ . What is the amount of oxygen molecules, in mol, that reacts? A. 4 B. 5 C. 6 D. 7 3. Under which combination of conditions is 1 mol of an ideal gas present? Volume Pressure ...

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The molar mass of a compound is approximately  $56 \text{ g mol}^{-1}$ . Which formula is possible for this Chemistry SL paper 1 TZ1 2009 - Exam - IB relics When  $100 \text{ cm}^3$  of  $1.0 \text{ mol dm}^{-3}$  HCl is mixed with  $100 \text{ cm}^3$  of  $1.0 \text{ mol dm}^{-3}$  NaOH, the temperature of the resulting solution increases by  $5.0^\circ \text{C}$ .

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